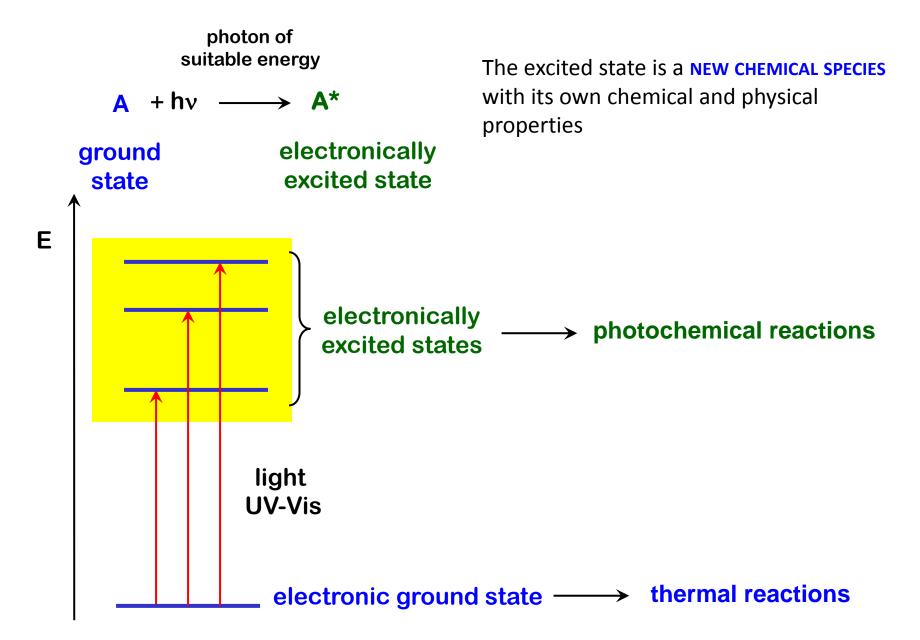
# **UV-Vis Absorption Spectra**

#### **ELECTRONIC EXCITED STATES**



### **Radiative transitions**

### **Absorption**

$$\Psi_i + h\nu \rightarrow \Psi_f$$

necessary but not sufficient condition:

$$h v = E_f - E_i = \Delta E$$

transition moment

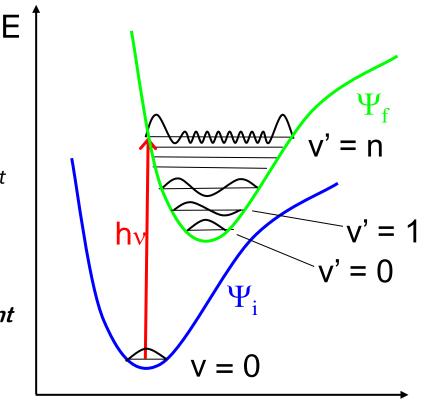
 $_{/}$   $P_{abs} \propto (TM)^2$ 

absorption probability

$$TM = \int \Psi_i \ \mu \Psi_f d au \quad \mu$$
 = electric dipole moment 
$$\Psi(Q,q,s) = \Lambda(Q)\Theta(q)S(s)$$

$$TM = \int \Psi_i \, \mu \Psi_f d\tau = \int (\Lambda_i \Theta_i S_i) \mu (\Lambda_f \Theta_f S_f) dQ dq ds$$

$$TM = \int \Lambda_i \Lambda_f dQ \int \Theta_i \mu \Theta_f dq \int S_i S_f ds$$



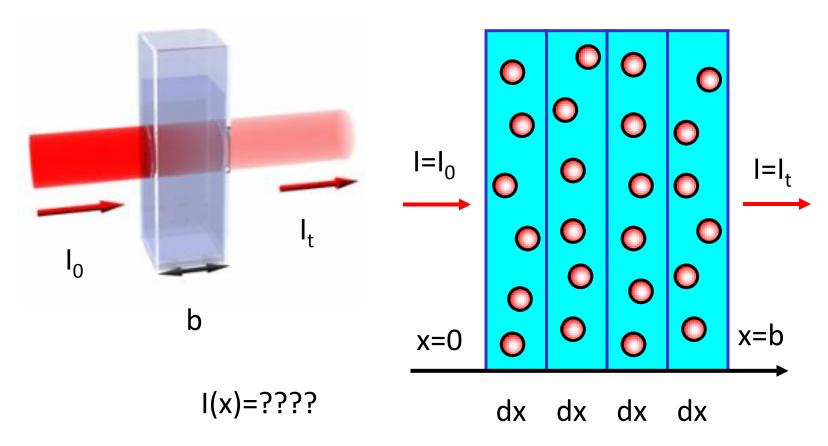
Franck-Condon factor

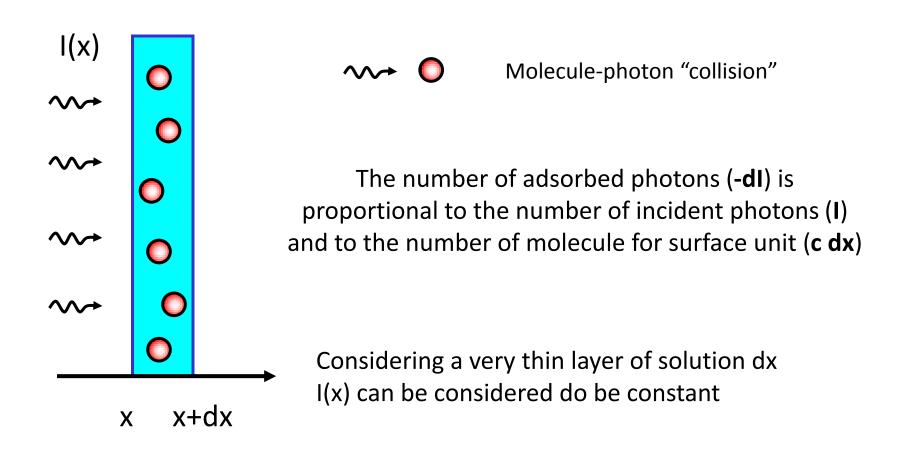
is large only 
$$\int \Lambda_i \Lambda_f dQ$$
 for v=0 and  $v'=n$ 

### Legge di Lambert-Beer

Light intensity decrease inside the sample from I<sub>0</sub> to I<sub>t</sub>

Molar concentration= c





Introducing the molar absorption coefficient arepsilon

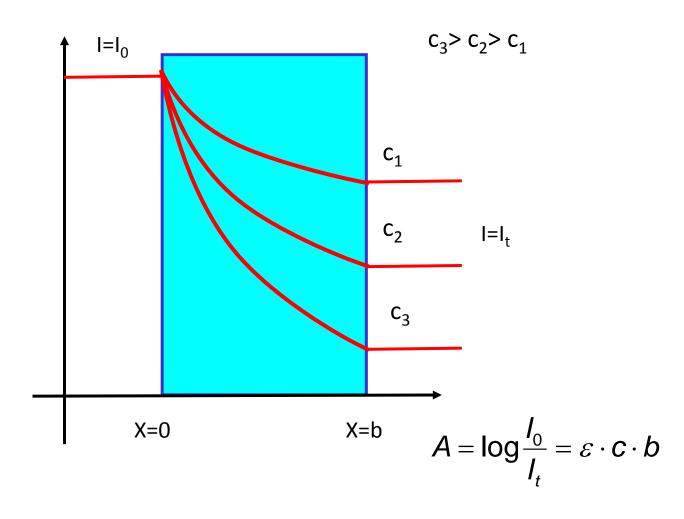
$$dI \propto -I \cdot c \cdot dx$$

$$\frac{1}{I}dI \propto -c \cdot dx \qquad \int_{I_o}^{I_t} \frac{1}{I} dI \propto -\int_{0}^{b} c \cdot dx$$

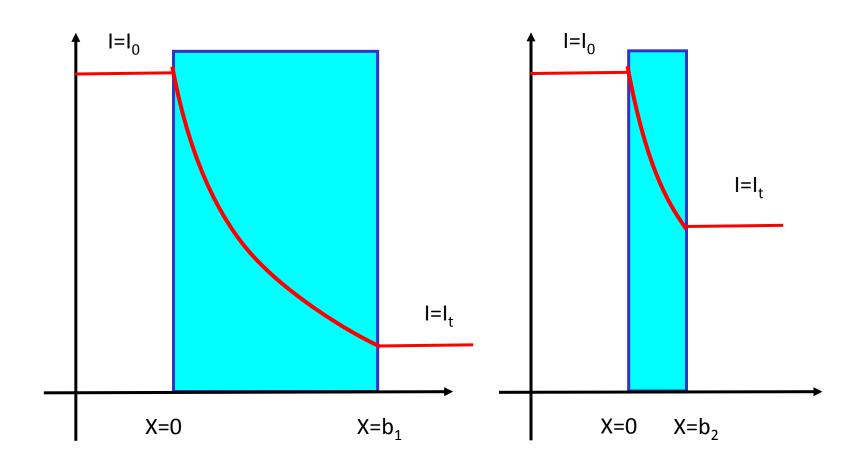
$$A = \log \frac{I_0}{I_t} = \varepsilon \cdot c \cdot b$$
 log is decimal logarithm!

$$I_t = I_0 \cdot 10^{-A} = I_0 \cdot 10^{-\varepsilon \cdot c \cdot b}$$
$$I(x) = I_0 \cdot 10^{-\varepsilon \cdot c \cdot x}$$

Effect of concentration

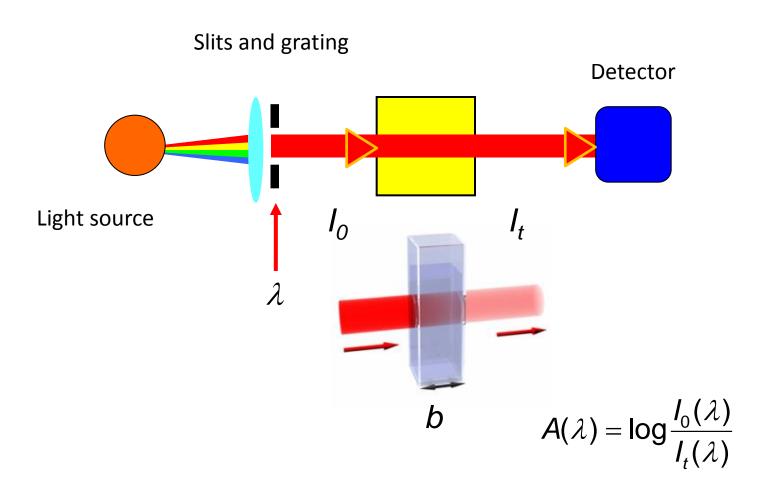


Effect of the optical path



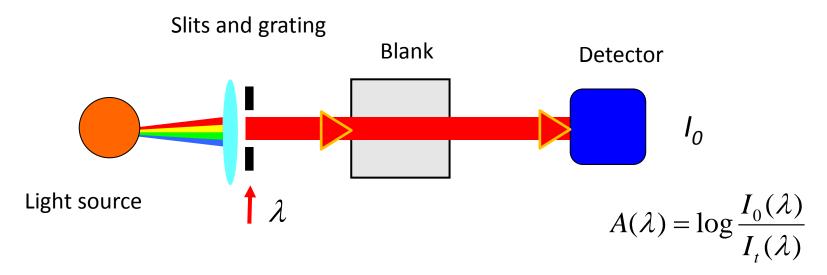
### **SPECTROPHOTOMETER: SINGLE BEAM**

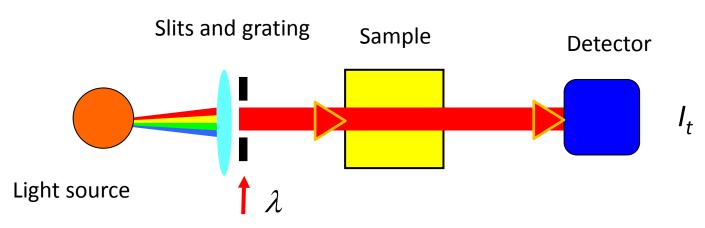
(UV-VIS ABSORPTION SPECTRA)



### **SPECTROPHOTOMETER: SINGLE BEAM**

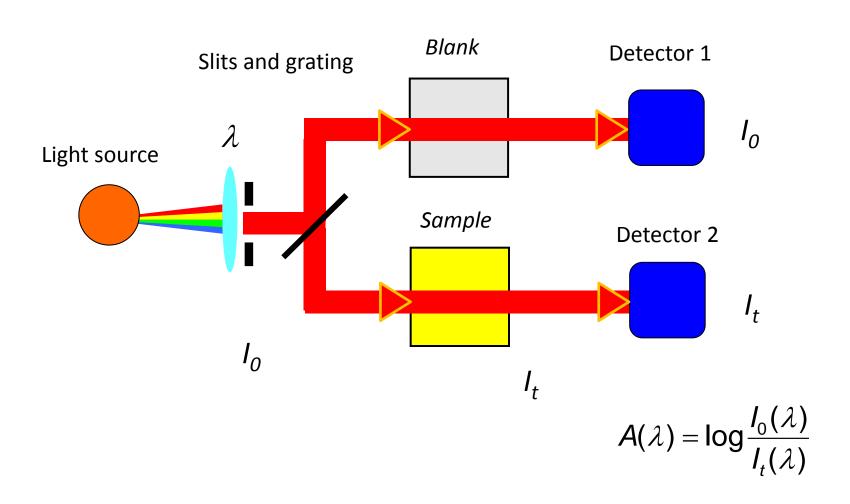
 $I_0$  and  $I_t$  are measured at different times



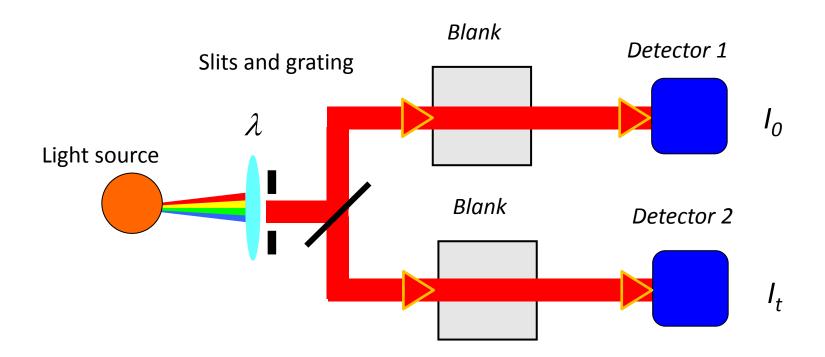


### **SPECTROPHOTOMETER: DOUBLE BEAM**

 $I_0$  and  $I_t$  are measured simultaneously



### **BASELINE (ZERO)**



- -The absorption spectrum is measured blank against blank or air against air
- -The spectrum obtained is automatically subtracted from the new spectra

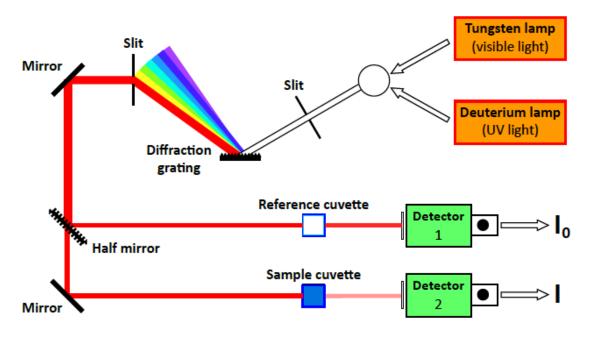
### What A values can be measured?

Maximum value: the instrument measures the current of each detector (1 and 2). This current is greater than zero even if the sample completely absorbs the incident light (dark current)

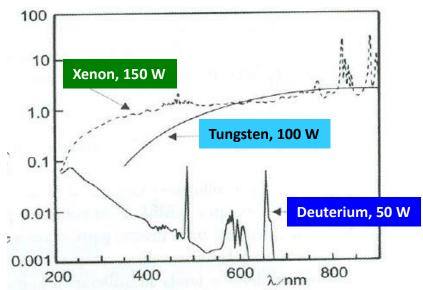
$$A(\lambda) = \log \frac{I_0(\lambda)}{I_t(\lambda)} \qquad A(\lambda) = \log \frac{i_1 + i_{1,buio}}{i_2 + i_{2,buio}} \quad A(\lambda) = \log \frac{i_1 + i_{1,buio}}{i_{2,buio}}$$

Minimum value: the stability of the zero line is not absolute and the variation represents the minimum limit for measurement (typically 0.001, in very well controlled conditions).

### **LIGHT SOURCES**



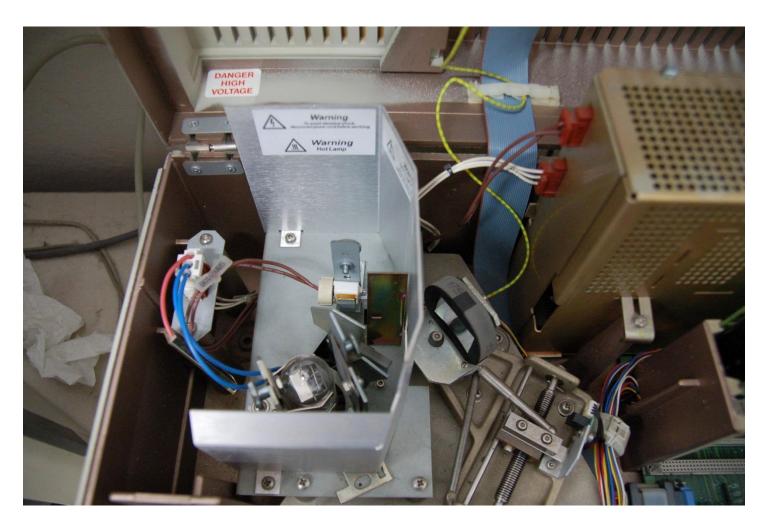




# **PERKIN ELMER Spectrophotometer**



# Lamps



# **Grating and beam splitter**



# Sample holder and detector



#### Parameters to set

**Spectral range:**  $\lambda_{max}$ ,  $\lambda_{min}$  in nm

Choise is based on the properties of the chromophore and the solvent

Scan speed: in nm /min

Increasing the speed worsens the resolution of the spectrum Increase the noise
In the same spectral range decreases the acquisition time

### Interval between successive points: in nm

Increasing the range decreases the resolution

Slit: in nm

It is expressed as a passband Increasing the slit decreases the resolution

### **SOFTWARE: UVWINLAB**

